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Engineering and Physical Sciences
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Organic Photovoltaic Solar Cells

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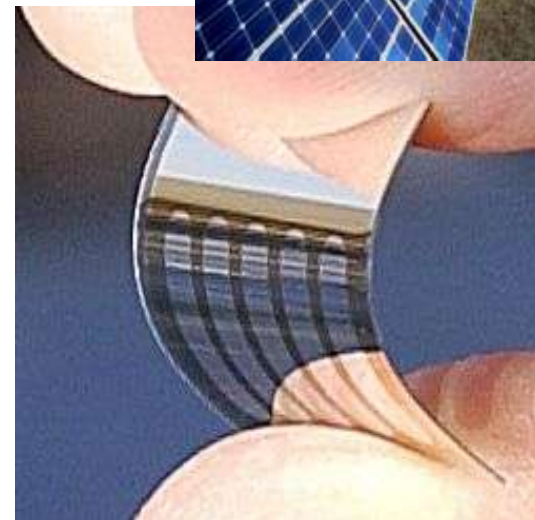


E-Futures



Outline

- Solar Power
- Photovoltaic Technology
- Organic Photovoltaics
 - Introduction
 - Cell Assembly
 - Cyclic Voltammetry
 - Dopants
 - Lifetime
 - Future
- Sheffield Solar Farm





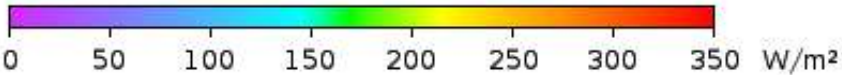
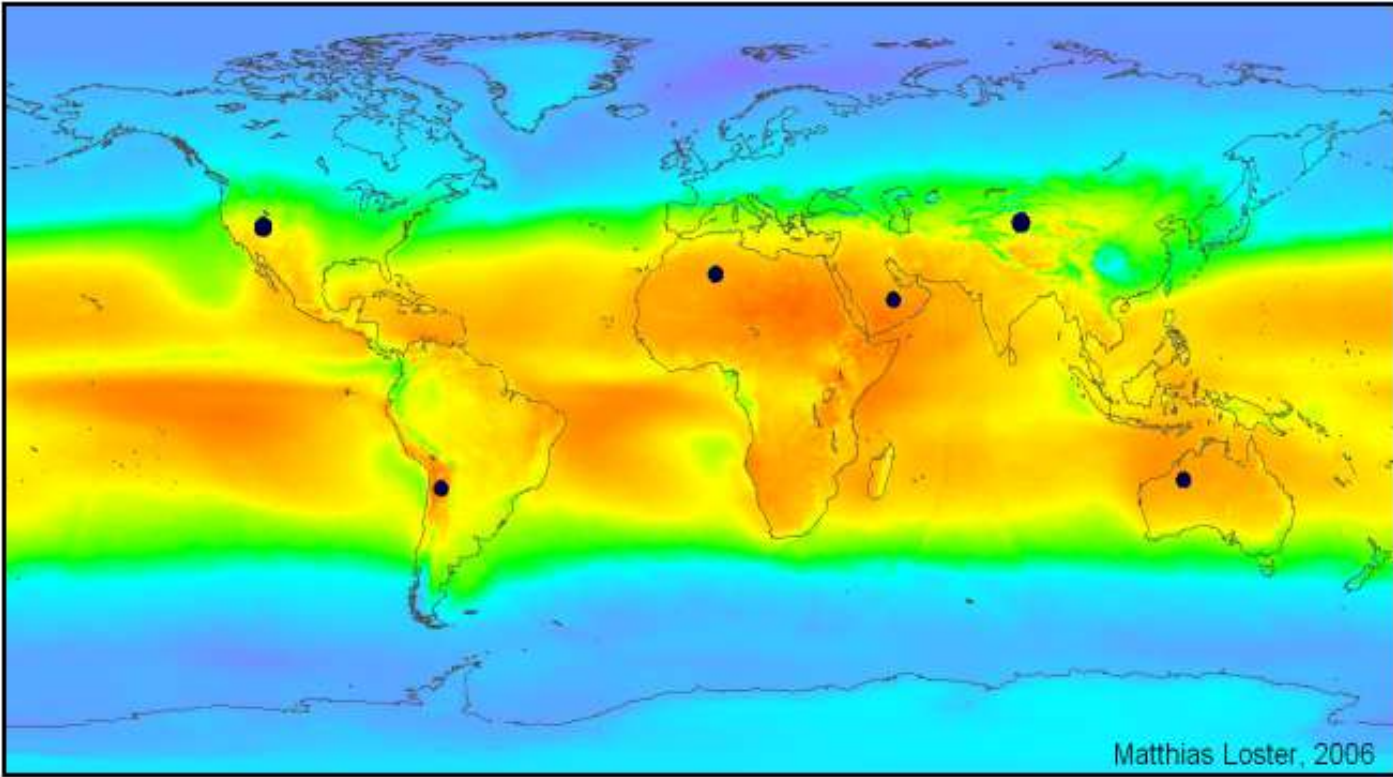
Solar Power

- The sun provides most of the energy required to drive the Earth's natural systems.
- Most sources of energy come directly or indirectly from the sun
- The amount of energy intercepted from the sun at the outer limits of the atmosphere is 1367 W/m^2
- Due to clouds, particles, water vapour and pollution, only part of this will reach the Earth's Surface



Solar Power

E-Futures



$\Sigma \bullet = 18 \text{ TWe}$



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Photovoltaic Technology

- Photovoltaic technology is a potential solution for decreasing the world's reliance on fossil fuels



usgreenenergycenter.com

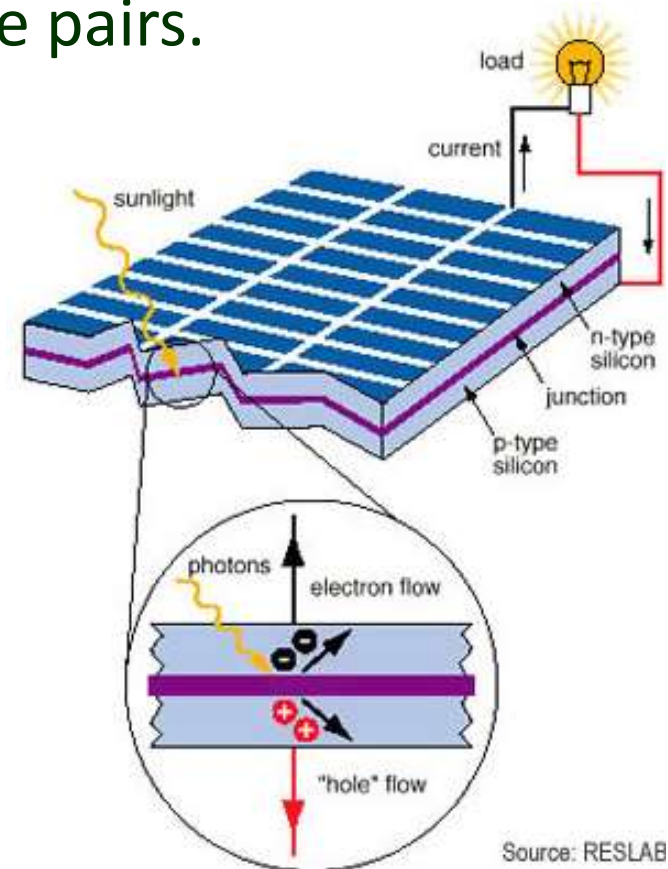
- Photovoltaics provide an efficient way to convert the energy from the sun directly into electricity



www.bp.com

Photovoltaic Technology

- When photons from the sun hit a cell, they are absorbed and create electron-hole pairs.
- The electrons and holes are attracted to electrodes at opposite surfaces of the semiconductor, creating a potential difference, and electricity can be generated



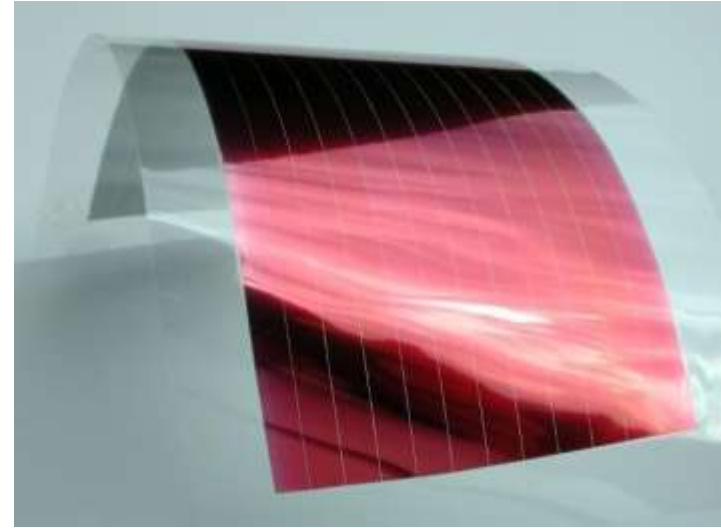


Introduction to OPVs

Traditional solar panels are energy-intensive and costly to manufacture.

Pros

- Organic polymers cost-effective, flexible and lightweight
- Can be solution-processed – ultimate aim is roll-to-roll fabrication



Cons

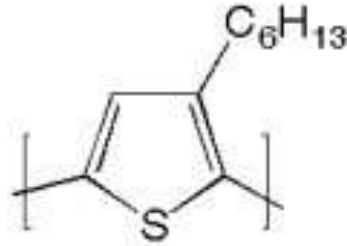
- Lifetime of OPVs is not yet established
- Need to increase speed of charge generation and extraction



Materials

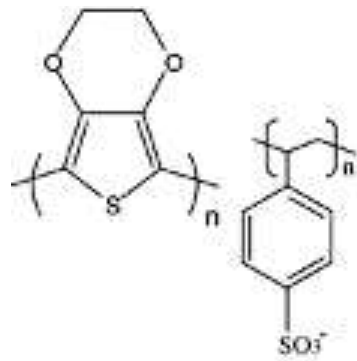
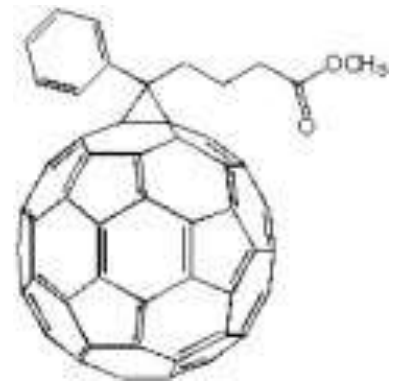
Materials commonly used in OPVs are:

- P3HT –electron donor, hole-transport material
- PCBM – electron acceptor, a fullerene derivative
- PEDOT:PSS – hole-transport (electron-blocking) layer



P3HT Poly(3-hexylthiophene)

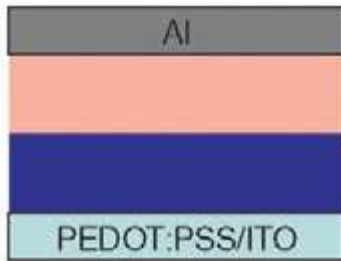
PCBM [6,6]-phenyl-C61-butyric acid methyl ester



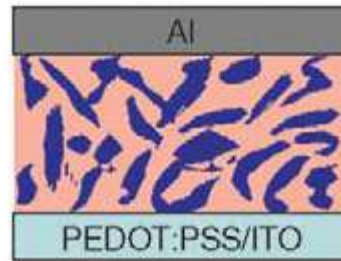
PEDOT:PSS Poly(3,4-ethylenedioxythiophene) : Poly(styrenesulphonate)



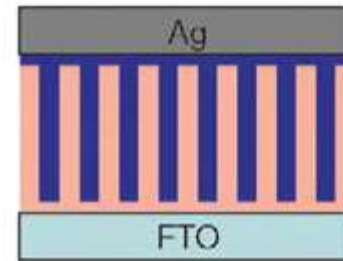
Junctions



Planar



Bulk



Ordered

The distance between P3HT and PCBM is crucial to achieving a current.

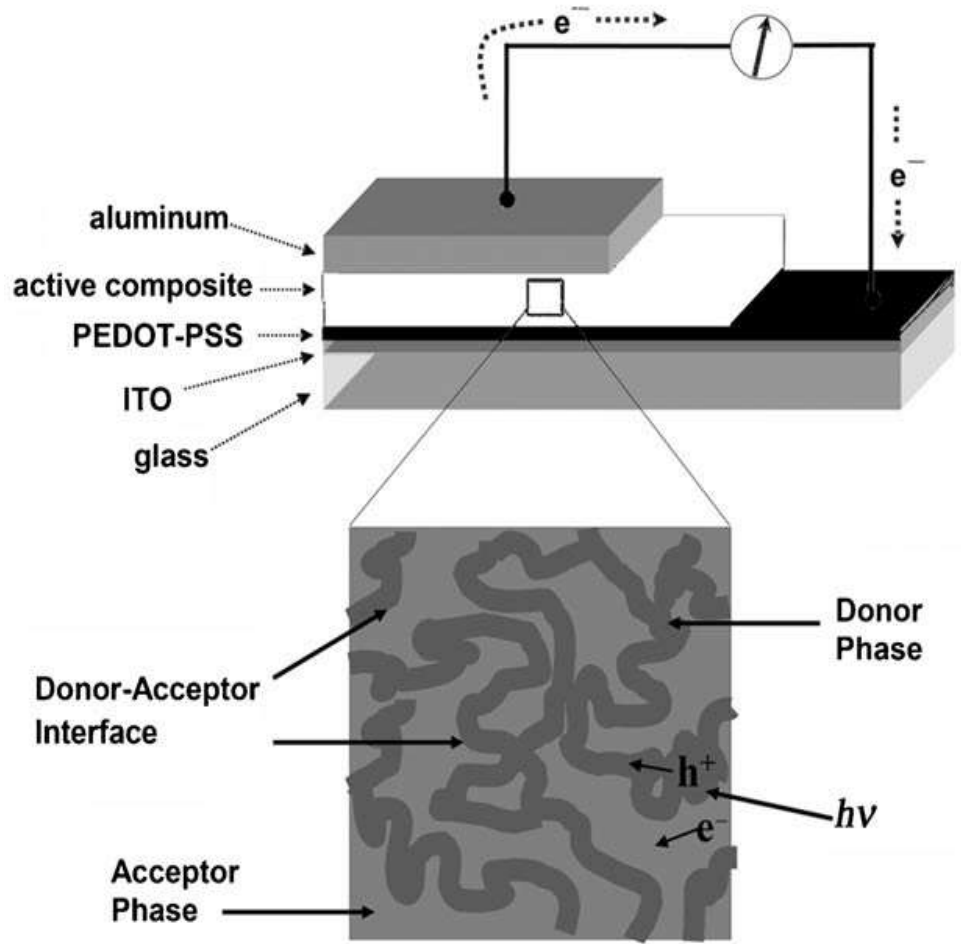
A blended composite of P3HT and PCBM is used to create a bulk heterojunction.

The two components phase-segregate and allow charge transport pathways between electrodes.



Cell Assembly

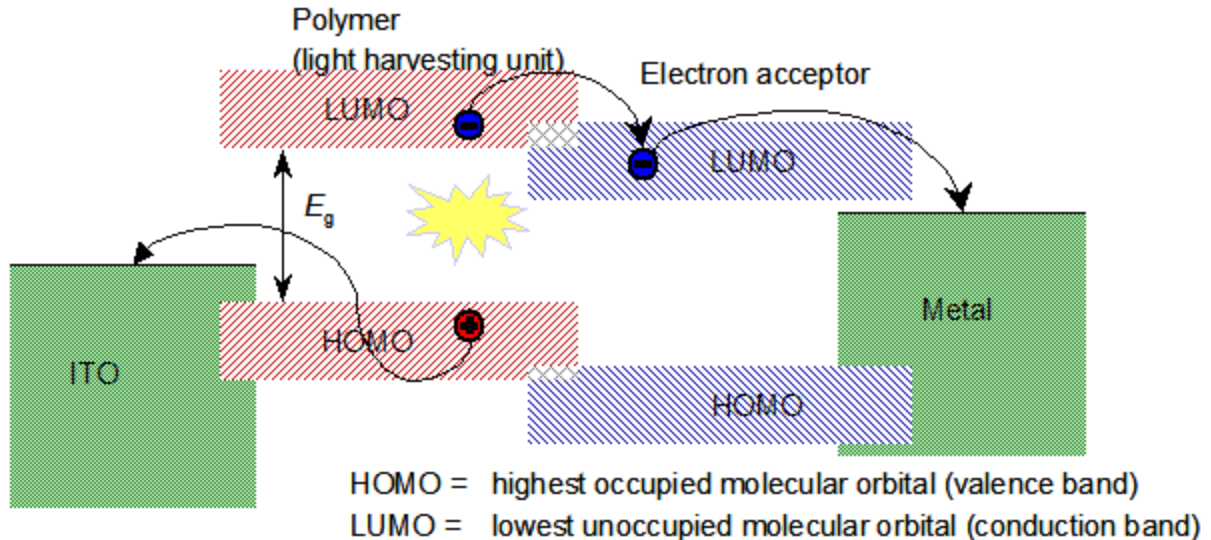
1. Remove protective photoresist from ITO-covered glass substrate.
2. Spin-coat with hole-transport layer (usually PEDOT:PSS, ~70 nm). Thermal anneal.
3. Spin-coat with active layer (P3HT:PCBM, 100-300 nm). Solvent anneal.
4. Deposit Aluminium cathode by evaporation.
5. Encapsulation.





Dopant Effect

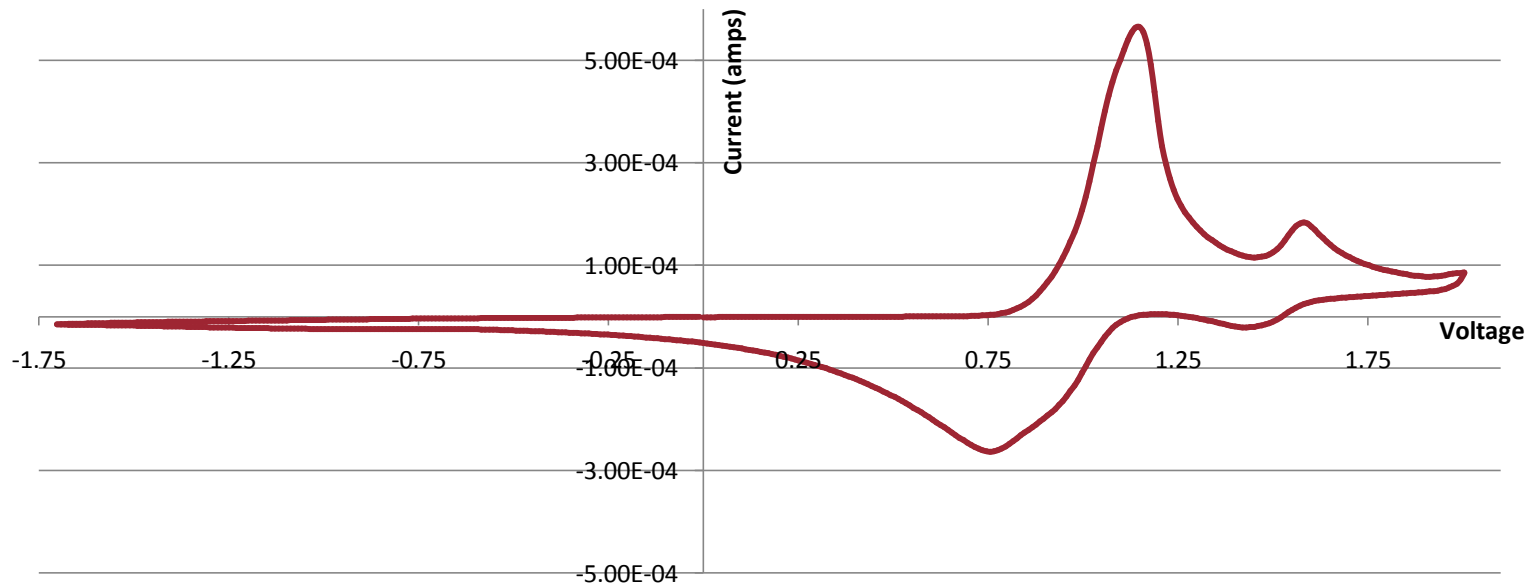
- Not always understood – conductivity increased accidentally with impurities!
- Fermi level – electronic equilibrium between materials in the cell.
- Dopant acts as electron acceptor.



Cyclic Voltammetry

- Shows oxidation and reduction of compound
- Voltage at which oxidation occurs in polymer and dopant can be compared

**BFE (four layers of film from Toluene) in MeCN (0.2 M TBAPF6)
vs. Ag/AgCl, 200 mV/s**

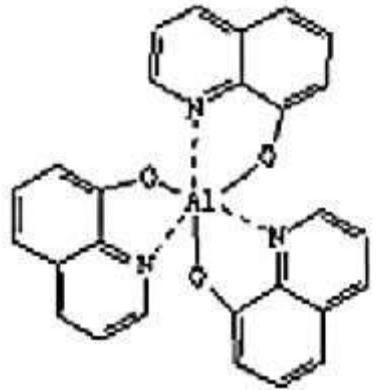




Current Dopants

Metals

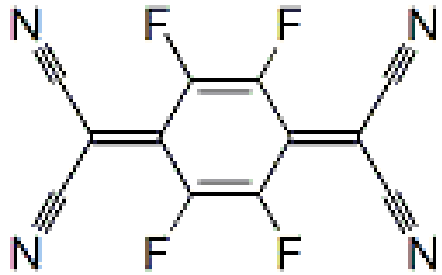
- Oxides: WO_3 , MoO_3
- Alq_3
- insoluble
- vacuum deposition



Alq₃

Organics

- F4-TCNQ
- F2-HCNQ
- triarylamine derivatives
- volatile



F4-TCNQ

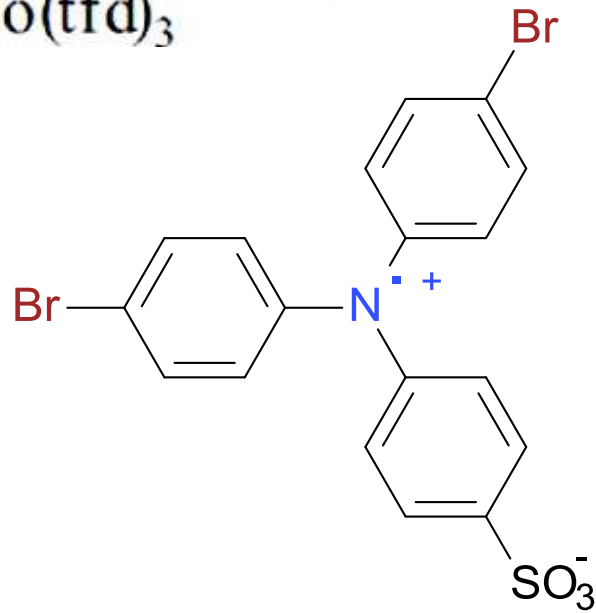
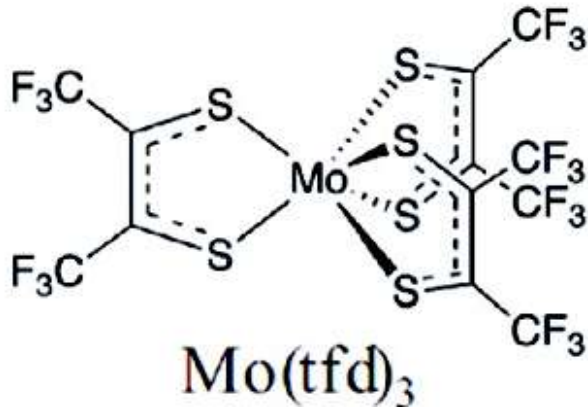


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Future Work

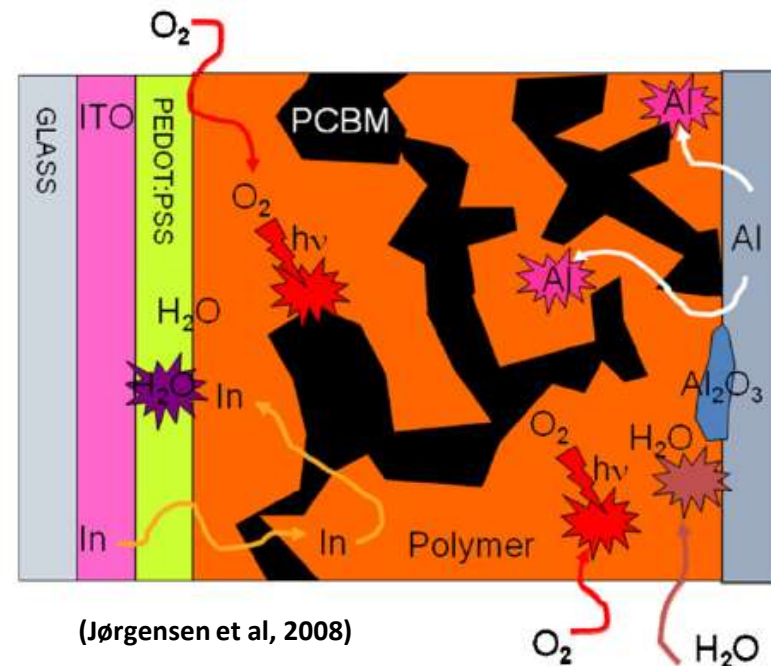
Dopant requirements:

- Have widely controllable solubility
- Have potential to dope a range of polymeric semiconductors
- Should not react with the host polymer
- Neutral agent, incorporated as an inert, immobile counter ion in the film



Lifetime/Stability

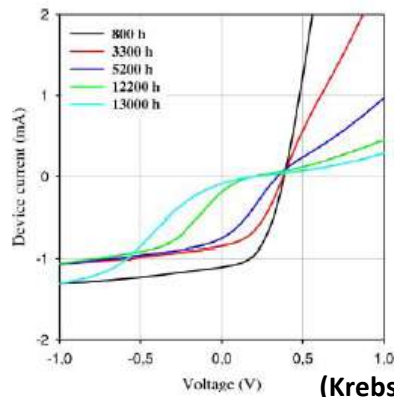
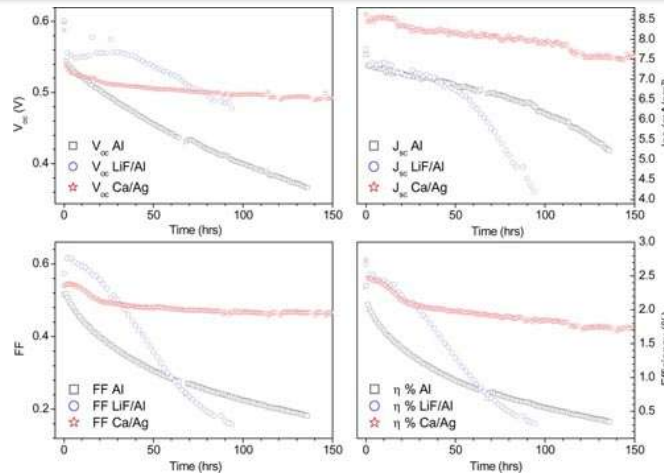
- Unfortunately OPVs currently have very short lifetimes
- Several possibilities for mechanism of degradation
 - Environmental and intrinsic factors
 - Light (photodegradation)
 - Temperature, humidity, presence of oxygen
 - Chemical composition, morphology
- Can be physical or chemical degradation





Mechanisms for Degradation

- Composition of cathode
 - Reduction in I_{sc} over time
- Thermal Degradation of PEDOT:PSS layer
 - Irreversible structural changes
- Light induced degradation of active layer
 - Changes to UV-Visible spectrum and IR spectra
- Diffusion of oxygen and water



(Krebs, Norrman, 2007)

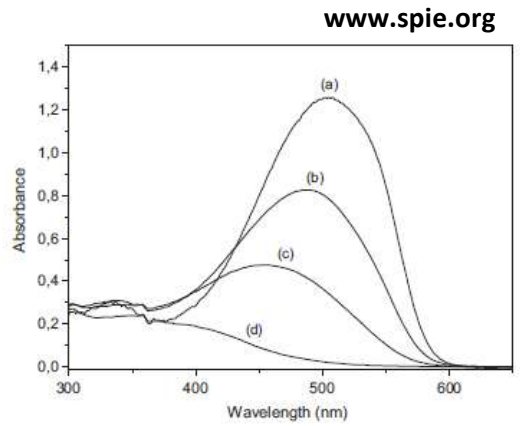


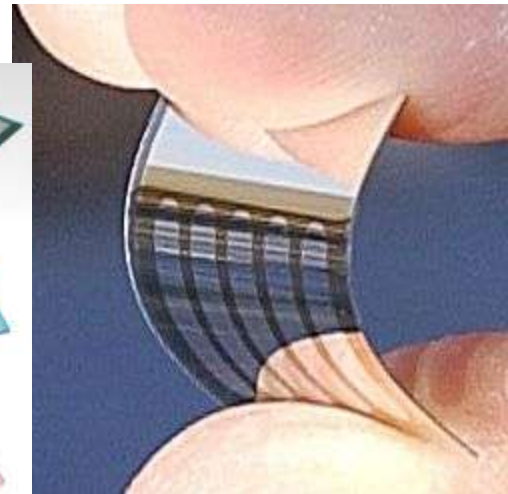
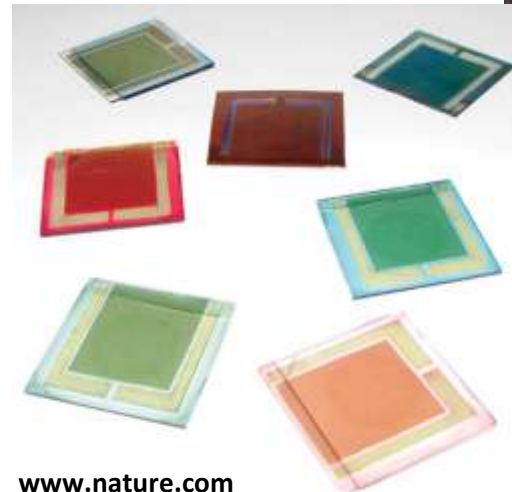
Fig. 1 Changes in the UV-visible spectrum of an MDMO-PPV thin film during photo-oxidation in the SEPAP 12/24 device: (a) 0 min; (b) 45 min; (c) 75 min; (d) 120 min.

(Rivaton et al, 2010)



The Future

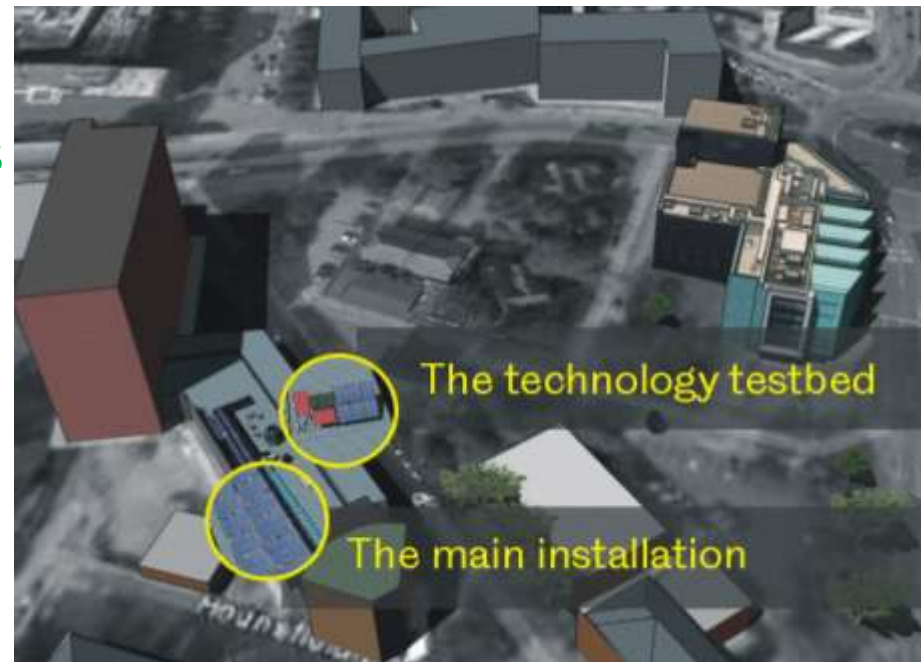
- OPV devices offer potential for commercialisation of solar cells
- Potential for large scale, low cost production
- Research efforts in:
 - Colour for building integration
 - Flexibility
 - Transparency
 - Increasing efficiency
 - Increasing stability
- Real-world testing





The Sheffield Solar Farm

- Primarily a 70m² silicon PV installation plus a technology testbed
- Comparison of new commercial and academic technologies and with standard laboratory based tests of efficiencies and lifetimes
- Lightweight flexible thin film photovoltaics for such as amorphous silicon, cadmium telluride and CIGS
- Dye-sensitised solar cells (Graetzel cells)
- Polymer photovoltaics, Bio-organic hybrid devices, and compound semiconductor devices



The Sheffield Solar Farm

- Pyranometer installed to provide live feed of both direct and diffuse radiation
- Allows analysis of performance related to actual ambient conditions



- Promotion of knowledge Transfer
- Real world testing





References

- Mayer, A.C., et al., Polymer-based solar cells. *Materials Today*, 2007. 10(11): p. 28-33.
- Thompson, B.C. and J.M.J. Frechet, Organic photovoltaics - Polymer-fullerene composite solar cells. *Angewandte Chemie-International Edition*, 2008. 47(1): p. 58-77.
- Kao, P.C., et al., Improved efficiency of organic photovoltaic cells using tris (8-hydroxy-quinoline) aluminum as a doping material. *Thin Solid Films*, 2009. 517(17): p. 5301-5304.
- Qi, Y.B., et al., A Molybdenum Dithiolene Complex as p-Dopant for Hole-Transport Materials: A Multitechnique Experimental and Theoretical Investigation. *Chemistry of Materials*, 2010. 22(2): p. 524-531.
- Gao, Z.Q., et al., An organic p-type dopant with high thermal stability for an organic semiconductor. *Chemical Communications*, 2008(1): p. 117-119.
- S. A. Gevorgyan, M. Jorgensen, F. C. Krebs. (2008). "A setup for studying stability and degradation of polymer solar cells." *Solar Energy Materials & Solar Cells* **92**: 736-745
- F. C. Krebs, K. Norrman. (2007). "Analysis of the Failure Mechanism for a Stable Organic Photovoltaic During 10000 h of Testing." *Progress in Photovoltaics: Research and Applications* **15**: 697-712
- M. Jorgensen, K. Norrman, F. Krebs. (2008). "Stability/degradation of polymer solar cells." *Solar Energy Materials & Solar Cells* **92**: 686-714
- A. Rivaton, S. Chambon, M. Manceau, J-L. Gardette, N. Lemaitre, S. Guillerez. (2010). "Light-induced degradation of the active layer of polymer-based solar cells." *Polymer Degradation and Stability* **95**: 278-284